

Application of a Sequential Multicomponent Assembly Process/ Huisgen Cycloaddition Strategy to the Preparation of Libraries of 1,2,3-Triazole-Fused 1,4-Benzodiazepines

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Supporting Information

ABSTRACT: A strategy featuring a multicomponent assembly process followed by an intramolecular azide-alkyne dipolar (Huisgen) cycloaddition was implemented for the facile synthesis of three different 1,2,3-triazolo-1,4-benzodiazepine scaffolds. A diverse library of 170 compounds derived from these scaffolds was then created through N-functionalizations, palladium-catalyzed cross-coupling reactions, and several applications of α -aminonitrile chemistry.

KEYWORDS: triazole, dipolar cycloaddition, azide—alkyne, benzodiazepine, multicomponent reaction, α -aminonitrile, diversity oriented synthesis

he identification of biologically active small molecules through high-throughput screening (HTS) is a common strategy in campaigns to discover both novel drug leads and molecular probes that may be used to study aberrant biological pathways leading to disease. Because of their high hit rates in HTS and "drug-like" pharmacological profiles, derivatives of privileged structures are uniquely well-suited as molecular scaffolds in drug discovery.² Straightforward variation of the substituents on privileged scaffolds often leads to potent and selective binders for multiple biological targets from a single library.

More than 20 years ago, Evans identified the benzodiazepine ring system as the archetypal privileged structure, thereby introducing the term into medicinal chemistry.^{3,4} Compounds derived from the 1,4-benzodiazepine ring system bind to a multitude of targets, including G protein-coupled receptors (GPCRs), ligand-gated ion channels, and enzymes.^{2a} The propensity of compounds derived from the benzodiazepine nucleus to bind to GPCRs has been suggested to arise from the ability of the scaffold to act as a structural mimic of peptide β -turns⁵ and α -helices.⁶ Such secondary structural elements orient substituents in a manner that enhances protein binding. Indeed, virtually all GPCR-binding ligands adopt α -, β - or γ turns.⁷ It is therefore unsurprising that the 1,4-benzodiazepine substructure is a common structural subunit in numerous pharmaceutical agents, biological probes, and bioactive natural products, such as diazepam (1),8 devazepide (2),9 and anthramycin (3), 10 respectively (Figure 1).

Figure 1. Derivatives of the 1,4-benzodiazepine ring system.

A key consideration inherent in designing and preparing novel compound libraries for HTS is the efficient synthesis of core scaffolds that bear functionality to enable facile diversification. In this context, we have recently developed a Mannich-type multicomponent assembly process (MCAP) to construct aryl aminomethyl adducts that are suitably functionalized for use in subsequent cyclizations to afford a variety of nitrogen-containing heterocyclic structures.¹¹ This sequential MCAP/cyclization strategy has been successfully utilized to rapidly access natural products^{11,12,13a} and unnatural heterocyclic frameworks.^{13,14} We also applied this approach to the

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syntheses of diverse, medicinally relevant scaffolds based upon the 1,2,3-triazolo-1,4-benzodiazepine core 6 (Scheme 1), 15,16 2-aryl

Scheme 1. Synthesis of Scaffolds 6 and 7

Scheme 2. Nitrogen Functionalization of Scaffolds 6 and 7^a

a) **20**, Et₃N, CH₂Cl₂. b) **21**, Et₃N, CH₂Cl₂. c) **22**, CH₂Cl₂. d) **23**, CH₂Cl₂. e) **24**, Et₃N, CH₂Cl₂. f) **25**, NaBH(OAc)₃, AcOH, DCE. g) **26**(1 -5}, Pd(OAc)₂ (5 mol%), (±)-BINAP (7.5 mol%), NaOt-Bu, toluene, 80 °C. h) **26**(6 }, Pd₂(dba)₃ (4 mol%), dppp (8 mol%), NaOt-Bu, toluene, 70 °C.

"Reaction conditions: (a) **20**, Et₃N, CH₂Cl₂; (b) **21**, Et₃N, CH₂Cl₂; (c) **22**, CH₂Cl₂; (d) **23**, CH₂Cl₂; (e) **24**, Et₃N, CH₂Cl₂; (f) **25**, NaB(OAc)₃H, AcOH, DCE; (g) **26**{1-5}, Pd(OAc)₂ (5 mol%), (±)-BINAP (7.5 mol %), NaOt-Bu, toluene, 80 °C; (h) **26**{6}, Pd₂(dba)₃ (4 mol %), dppp (8 mol%), NaOt-Bu, toluene, 70 °C.

piperidines, ^{15,17} tetrahydroisoquinolines, ^{15,18} as well as isoindolinones, norbenzomorphans, benzazocines, and benzoxacines. ^{15,19} We now report the application of this general approach for diversity oriented synthesis (DOS) to the preparation of libraries of 1,2,3-triazolo-1,4-benzodiazepines, a subclass of the benzodiazepine privileged structure whose biological properties have not been widely explored. ²⁰

As described previously, the parent 1,2,3-triazolo-1,4-benzodiazepine **6** and brominated analog 7 were readily prepared on multigram scale by a reductive amination and azide—alkyne dipolar (Huisgen) cycloaddition sequence (Scheme 1).¹⁵ This approach afforded scaffolds **6** and **7** in high purity after recrystallization and represents a considerably more expedient access to compound **6** than previously reported.²¹

Scheme 3. Cross-Coupling Reactions with Scaffold 7

Scheme 4. Two-Dimensional Library Derived from Scaffold π^a

a) **20**, Et₃N, CH₂Cl₂. b) **21**, Et₃N, CH₂Cl₂. c) **22**, CH₂Cl₂. d) **23**(3}, CH₂Cl₂. e) **25**, RCHO, NaBH(OAc)₃, AcOH, DCE. f) **26**(1), Pd(OAc)₂ (5 mol%), (±)-BINAP (7.5 mol%). NaOt-Bu, toluene. 80 °C.

"Reaction conditions: (a) **20**, Et_3N , CH_2Cl_2 ; (b) **21**, Et_3N , CH_2Cl_2 ; (c) **22**, CH_2Cl_2 ; (d) **23**{3}, CH_2Cl_2 ; (e) **25**, RCHO, $NaB(OAc)_3H$, AcOH, DCE; (f) **26**{1}, $Pd(OAc)_2$ (5 mol%), (\pm)-BINAP (7.5 mol%), NaOt-Bu, toluene, 80 °C.

The two heterocycles 6 and 7 were then subjected to various tactics for N-functionalization to diversify the scaffolds into libraries of amides 8 and 9, sulfonamides 10 and 11, ureas 12 and 13, thioureas 14 and 15, carbamate 16, and amine derivatives 17–19 (Scheme 2, Figure 2, Table 1). Reagents were typically used in excess to mitigate for variations in the purity of reagents, which were used without purification, and reaction efficiency. Functionalizing agents were selected to contain a mixture of aliphatic and aromatic groups. Different aliphatic substituents were chosen to vary the nonpolar surface area, whereas aromatic

Scheme 5. Synthesis of Nitrile-Substituted Scaffold 44

Figure 2. Nitrogen-functionalizing agents used to derivatize scaffolds 6 and 7.

22{8}

22{4}

NCO

22{5}

NCO

ÒМе

22{9}

22{6}

groups were varied by selecting substituents that would modulate the electrostatic properties of the aromatic ring. In this manner, it would be possible to acquire preliminary structure activity

22{3}

22{1}

22{2}

NCO

22{7}

relationship (SAR) data from a small set of compounds. N-Arylation procedures to access chemset 19 were explored for the formation of p-toluidine derivative 19{2}. We discovered that

26{2}

26{3}

26{4}

26{1}

26{*5*}

26{6}

Table 1. Nitrogen Functionalization of Scaffolds 6 and 7

entry	scaffold	functionalizing agent	product	yield (%)	entry	scaffold	functionalizing agent	product	yield (%)
1	6	20 {1}	8{1}	95	37	7	22 {3}	13{3}	55
2	6	20 {2}	8{2}	96	38	7	22{4}	13{4}	quant.
3	6	20 {3}	8 {3}	quant.	39	6	23{1}	14 {1}	99
4	6	20 {4}	8{4}	93	40	6	23 {2}	14{2}	98
5	6	20 {5}	8 {5}	98	41	6	23 {3}	14{3}	98
6	6	20 { <i>6</i> }	8 {6}	96	42	6	23{4}	14{4}	83
7	6	20 {7}	8 {7}	98	43	6	23 {5}	14 {5}	94
8	6	20 {8}	8 {8}	98	44	6	23 {6}	14 { <i>6</i> }	95
9	6	20 {9}	8 {9}	98	45	6	23 {7}	14{7}	90
10	6	20 {10}	8 {10}	$82^{a,b}$	46	7	23 {1}	15 {1}	79
11	6	20 {11}	8{11}	95	47	7	23 {3}	15 {3}	83
12	7	20 {9}	9 {9}	90	48	7	23 {7}	15{7}	quant.
13	7	20 {12}	9{12}	67	49	6	24 {1}	16 {1}	88
14	6	21 {1}	10 {1}	97	50	6	24 {2}	16{2}	77
15	6	21{2}	10{2}	64	51	6	24 {3}	16 {3}	98
16	6	21{3}	10 {3}	91	52	6	25 {1}	17{1}	71
17	6	21{4}	10{4}	98	53	6	25 {2}	17{2}	quant.
18	6	21{5}	10 {5}	98	54	6	25 {3}	17{3}	98
19	6	21{6}	10 { <i>6</i> }	89	55	6	25{4}	17{4}	83
20	6	21{7}	10 {7}	94	56	6	25 {5}	17{5}	quant.
21	6	21{8}	10 {8}	88	57	6	25 {6}	17{6}	82
22	6	21{9}	10 {9}	81	58	6	25 {7}	17{7}	65
23	6	21 {10}	10 {10}	84	59	6	25 {8}	17{8}	60
24	7	21{1}	11 { <i>1</i> }	85	60	6	25 {9}	17{9}	96
25	7	21{4}	11{4}	53	61	6	25 {10}	17{10}	96
26	7	21{6}	11{6}	55	62	6	25 {11}	17{11}	56 ^c
27	6	22 {1}	12 { <i>1</i> }	99	63	7	25 {5}	18 {5}	98
28	6	22 {2}	12 {2}	79	64	7	25 {12}	18{12}	55
29	6	22 {3}	12 {3}	96	65	7	25 {13}	18{13}	70
30	6	22{4}	12{4}	99	66	6	26 {1}	19 {1}	96
31	6	22 {5}	12 {5}	99	67	6	26 {2}	19{2}	83
32	6	22 {6}	12 { <i>6</i> }	86	68	6	26 {3}	19 {3}	73
33	6	22 {7}	12 {7}	91	69	6	26{4}	19{4}	48
34	6	22{8}	12{8}	quant.	70	6	26 {5}	19 {5}	0
35	6	22{9}	12{9}	quant.	71	6	26 {6}	19 { <i>6</i> }	85 ^c
36	7	22 {2}	13{2}	58					

 $[^]a$ An additional 3.0 equiv of Et $_3$ N were employed. b Isolated as the free-base. c No AcOH added.

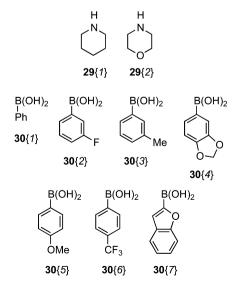


Figure 3. Cross-coupling partners employed with scaffold 7.

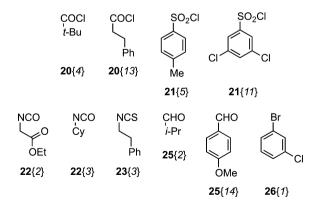


Figure 4. Nitrogen-functionalizing agents used in the two-dimensional library derived from scaffold 7.

conditions reported by Buchwald utilizing $Pd(OAc)_2$, (\pm) -BINAP, and NaOt-Bu were superior to alternative protocols in which $Pd_2(dba)_3$ was employed as the Pd source.²² Use of this method

allowed the introduction of aryl halides onto the parent scaffold (19{1} and 19{3}) without the intervention of observable side reactions involving secondary cross-couplings. The 2-amino pyridine derivative $19\{6\}$ was also prepared using a procedure described by Buchwald, but a catalyst loading of 4 mol % Pd_2dba_3 was necessary for complete conversion.²³

The aryl bromide functionality present within scaffold 7 was exploited as a handle for diversification through several palladium-catalyzed cross-couplings (Scheme 3, Figure 3, Table 2). To avoid homocoupling of 7 during Buchwald—

Table 2. Cross-Coupling Reactions with Scaffold 7

	1:	1 ,	. 11 (0/)
entry	coupling partner	product	yield (%)
1	29 {1}	2 7{1}	75
2	29 {2}	27 {2}	69
3	30 {1}	28 {1}	84
4	30 {2}	28 {2}	73
5	30 {3}	28 {3}	74
6	30 {4}	28{4}	78
7	30 {5}	28 {5}	78
8	30 { <i>6</i> }	28 { <i>6</i> }	76
9	30 {7}	28 {7}	92
3 4 5 6 7 8	30{1} 30{2} 30{3} 30{4} 30{5} 30{6}	28{1} 28{2} 28{3} 28{4} 28{5} 28{6}	84 73 74 78 78 76

Hartwig aminations, it was necessary to employ a 10-fold excess of amines 29 to afford the aniline derivatives 27 (Figure 3). Suzuki cross-couplings of 7 with boronic acids 30 afforded biaryl products 28 bearing substituents having a broad range of electronic properties. The use of heteroaromatic boronic acid derivatives is exemplified by the synthesis of 28{7}.

Five cross-coupled products $(27\{1-2\}, 28\{1-2\}, \text{ and } 28\{5\})$ were subjected to ten independent N-functionalizations to afford a "two-dimensionally" functionalized collection of 50 compounds (chemsets 31-42) in generally high yields (Scheme 4, Figure 4, and Table 3).

Because compounds bearing the α -aminonitrile functionality are synthetically versatile intermediates, we envisioned that a scaffold containing this subunit might facilitate diversification through a number of strategies.²⁴ The Strecker MCAP employing benzaldehyde 4 and propargylamine furnished the α -aminonitrile 43 in 78% yield (Scheme 5). Upon mild heating, 43 underwent facile Huisgen cycloaddition to deliver the cyanosubstituted scaffold 44 in 88% yield. It was necessary to perform this reaction under dilute conditions and at a temperature not exceeding 60 °C to avoid elimination of HCN from the product 44. Interestingly, when 43 was N-acylated, the intermediate amide underwent facile cycloaddition at ambient temperature to yield the amide derivative of the scaffold $45\{1\}$. This observation led to the development of a convenient, 4-component process to afford 45{1} directly from benzaldehyde 4 in 49% yield (Scheme 5). This MCAP approach is an expedient method to access amide derivatives of 44. However, N-functionalization of this scaffold is a better strategy for the library synthesis outlined in Scheme 6, because this approach allows the use of a number of parallel diversification reactions to give products that may be easily purified.

N-Acylations and N-sulfonylations of scaffold 44 to produce chemsets 45 and 46 were conducted using pyridine as the base; when triethylamine was employed, elimination of HCN was a significant side-reaction (Scheme 6, Figure 5, Table 4). Reductive amination of 44, which was achieved under the same conditions as those described for the preparation of 17 and 18 (Scheme 2), generally proceeded in good yields. However, the reaction of 44 with the electron poor aldehyde

Table 3. Two-Dimensional Library Derived from Scaffold 7

		,		
entry	amine input	functionalizing agent	product	yield (%)
1	2 7{1}	20 {4}	31{1,4}	quant.
2	2 7{1}	20 {13}	31 {1,13}	86
3	2 7{1}	21 {5}	33 {1,5}	86
4	27 {1}	21 {11}	33{1,11}	87
5	27 {1}	22 {2}	35 {1,2}	92
6	27 {1}	22 {3}	35 {1,3}	7
7	27 {1}	23 {3}	37{1,3}	91
8	27 {1}	25 {2}	39 {1,2}	78
9	27 {1}	25 {14}	39 {1,14}	88
10	2 7{1}	26 {1}	41 {1,1}	83
11	27 {2}	20 {4}	31{2,4}	94
12	27 {2}	20 {13}	31{2,13}	73
13	2 7{2}	21 {5}	33{2,5}	87
14	2 7{2}	21 {11}	33{2,11}	88
15	2 7{2}	22 {2}	35 {2,2}	76
16	2 7{2}	22 {3}	35 {2,3}	90
17	2 7{2}	23 {3}	3 7{2,3}	94
18	2 7{2}	25 {2}	39 {2,2}	88
19	2 7{2}	25 {14}	39 {2,14}	79
20	2 7{2}	26 {1}	41 {2,1}	90
21	28 {1}	20 {4}	32{1,4}	quant.
22	28 {1}	20 {13}	32 {1,13}	88
23	28 {1}	21 {5}	34{1,5}	91
24	28 {1}	21 {11}	34{1,11}	quant.
25	28 {1}	22 {2}	36{1,2}	86
26	28 {1}	22 {3}	36 {1,3}	97
27	28 {1}	23 {3}	38{1,3}	98
28	28 {1}	25 {2}	40 {1,2}	85
29	28 {1}	25 {14}	40 {1,14}	39
30	28 {1}	26 {1}	42 {1,1}	94
31	28 {2}	20 {4}	32 {2,4}	quant.
32	28 {2}	20 {13}	32 {2,13}	73
33	28 {2}	21{5}	34{2,5}	quant.
34	28 {2}	21 {11}	34{2,11}	quant.
35	28 {2}	22 {2}	36 {2,2}	90
36	28 {2}	22 {3}	36 {2,3}	quant.
37	28 {2}	23 {3}	38{2,3}	quant.
38	28 {2}	25 {2}	40 {2,2}	94
39	28 {2}	25{14}	40 {2,14}	35
40	28 {2}	26{1}	42 {2,1}	96
41	28 {5}	20{4}	32{5,4}	quant.
42	28 {5}	20 {13}	32 {5,13}	85
43	28 {5}	21{5}	34{5,5}	97
44	28 {5}	21{11}	34{5,11}	quant.
45	28 {5}	22{2}	36{5,2}	94
46	28 {5}	22 {3}	36 {5,3}	98
47	28 {5}	23 {3}	38{5,3}	99
48	28 {5}	25 {2}	40 {5,2}	94
49	28 {5}	25 {14}	40 {5,14}	40
50	28 {5}	26 {1}	42 {5,1}	99
-	(-)	,	C 9 3	

25{5} required the presence of trifluoroacetic acid and additional quantities of NaBH(OAc)₃ to consume the starting material 44 (Table 4, entry 19).

The acidic nature of the proton α to the cyano group in amide derivatives **45** was exploited to facilitate further elaboration of the scaffold (Scheme 7, Figure 6, Table 5). In the event, deprotonation of **45** with NaH, followed by trapping the intermediate carbanion with alkylating agents **50** delivered a small library of compounds bearing fully substituted carbon

Table 4. Nitrogen Functionalization of Scaffold 44

entry	functionalizing agent	product	yield (%)
1	20 {1}	45 {1}	92
2	20 {4}	45 {4}	92
3	20 {7}	45 {7}	93
4	20 {10}	45 {10}	92 ^a
5	20 {12}	45 {12}	49 ^b
6	20 {13}	45 {13}	95
7	20 {14}	45 {14}	98
8	20 {15}	45 {15}	66 ^c
9	20 {16}	45 {16}	quant.
10	20 {17}	45 {17}	quant.
11	21 {1}	46 {1}	97
12	21 {3}	46 {3}	91
13	21 {5}	46 {5}	86
14	21 {6}	46 { <i>6</i> }	90
15	21 {7}	46 {7}	91
16	21{9}	46 {9}	85
17	21 {11}	46 {11}	80
18	25 {1}	47 {1}	63
19	25 {5}	47 {5}	41 ^d
20	25 {12}	47 {12}	84
21	25 {14}	47 {14}	74
22	25 {15}	47 {15}	87
	,		

^aIsolated as the free-base. ^bYield after recrystallization from toluene (following chromatography). ^cYield after the product was washed with hexanes $(3 \times 5 \text{ mL})$ following chromatography. ^dThe addition of TFA (1.0 equiv) and extra NaBH(OAc)₃ (4.0 equiv) was required to fully consume 44.

Scheme 6. Nitrogen Functionalization of Scaffold 44

centers (chemset 48). Following literature precedent, vinyltriphenylphosphonium bromide (Schweizer's reagent) (51) was utilized as an electrophile to prepare a novel collection of pyrrolefused structures (chemset 49) through an addition/intramolecular Wittig reaction/elimination sequence. Notably, the Wittig reaction and aromatization steps were complete within 90 min at ambient temperature. With the sole exception of the pivalamide 45{4}, N-alkyl substituted amides were good substrates for this sequence of reactions. Aryl-substituted amides such as 45{15},

Scheme 7. Anion Derived Diversification of Amides 45

Scheme 8. Diversification of Aminonitriles 47 using the Bruylants Reaction

Table 5. Libraries Derived from Amides 45

entry	amide	electrophile	product	yield (%)
1	45 {1}	50 {1}	48 {1,1}	80
2	45 {1}	50 {2}	48 {1,2}	86
3	45 {15}	50 {1}	48 {15,1}	80
4	45 {15}	50 {2}	48 {15,2}	76
5	45 {1}	51	49 {1}	72
6	45 {13}	51	49 {13}	66
7	45 {4}	51	49 {4}	trace ^a
8	45 {15}	51	49 {15}	95
9	45 {16}	51	49 {16}	85
10	45 {7}	51	49 {7}	44
11	45 {10}	51	49 {10}	91
12	45 {17}	51	49 {17}	0

 a Trace formation of the pyrrole product $49\{4\}$ was observed in the 1 H NMR spectrum of the crude reaction mixture.

Table 6. Amines Derived from the Bruylants Reaction

entry	aminonitrile	organo-zinc precursor	product	yield (%)
1	47 {1}	54 {1}	52 {1,1}	73
2	47 {1}	54 {2}	52 {1,2}	86
3	47 {12}	54 {1}	52 {12,1}	75
4	47 {12}	54 {2}	52 {12,2}	73
5	47 {14}	54 {1}	52 {14,1}	80
6	47 {14}	54 {2}	52 {14,2}	81
7	47 {1}	55	53 {1}	74
8	47 {12}	55	53 {12}	77
9	47 {14}	55	53 {14}	75

Figure 5. Nitrogen functionalizing agents used with scaffold 44.

Mel BnBr
$$\bigwedge^{\bigoplus}$$
 PPh₃Bi

50 {1} 50 {2} 51

Figure 6. Electrophiles employed to trap anions derived from amides 45

45{16}, and 45{7} also participated in this transformation, as did the nicotinamide 45{10}, which represents the first example of the use of a heteroaryl-substituted amide in this reaction. On the other hand, treating the anion of 5-nitro-furanyl amide 45{17} with the vinyl phosphonium salt 51 formed a complex mixture of products; no pyrrole was observed. It should be noted that the bromo-substituted benzaldehyde 5 might be used as a starting material to prepare the corresponding bromo derivatives of 44, 45, 48, and 49. These compounds could then be further derivatized by palladium-catalyzed cross-coupling reactions to introduce aryl and amine substituents onto the benzodiazepine rings of 44, 45, 48, and 49.

We were also interested in exploiting the Bruylants reaction to generate substituted benzylamine derivatives **52** by displacing cyanide ion from α -aminonitriles **47** with organometallic reagents (Scheme 8, Figure 7, Table 6). After conducting a brief series of exploratory studies, we discovered that transmetalation of Grignard reagents to their corresponding organo-zinc species was necessary to avoid competitive deprotonation of the acidic proton α to the cyano group. Displacement of cyanide ion under 'Reformatsky-type' conditions with the enolate of ethyl acetate afforded chemset **53**; these compounds bear an ethyl ester group that might serve as a further point of diversification.

Figure 7. Precursors to organo-zinc reagents used in the Bruylants reaction.

In summary, a diverse collection of 170 1,2,3-triazolo-1, 4-benzodiazepines was prepared. A sequential MCAP/Huisgen cycloaddition strategy was instrumental for the rapid construction of the parent scaffolds. N-Functionalization and cross-coupling reactions were then applied to these scaffolds to enable the rapid production of derivatives through a combinatorial style approach, whereas exploitation of α -aminonitrile chemistry facilitated the introduction of significant structural diversity via a DOS approach. A complete Lipinski analysis of a representative selection of 80 compounds containing at least two examples of each structural subtype and having a maximal diversity of substituents was performed, and no compound was found that violated the Lipinski rule of five. ²⁹ Hence, the library members have physiochemical properties that are likely to be associated with good solubility, membrane permeability, and bioavailability. All compounds have been submitted to the National Institutes of Health (NIH) Molecular Libraries Small Molecule Repository (MLSMR) and are being distributed to the Molecular Libraries Probe Production Centers Network (MLPCN) for HTS in a wide range of biological assays. Selected compounds have also been sent to the National Institute of Mental Health, Psychoactive Drug Screening Program (NIMH PDSP). The further development and application of our

sequential MCAP/cyclization strategy to the synthesis of medicinally relevant small molecules is ongoing and will be reported in due course.

ASSOCIATED CONTENT

S Supporting Information

Experimental procedures, full characterization data for representative compounds, LCMS data for representative compounds, and tabulated Lipinski's rule parameters for representative compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

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